

Alloying Elements

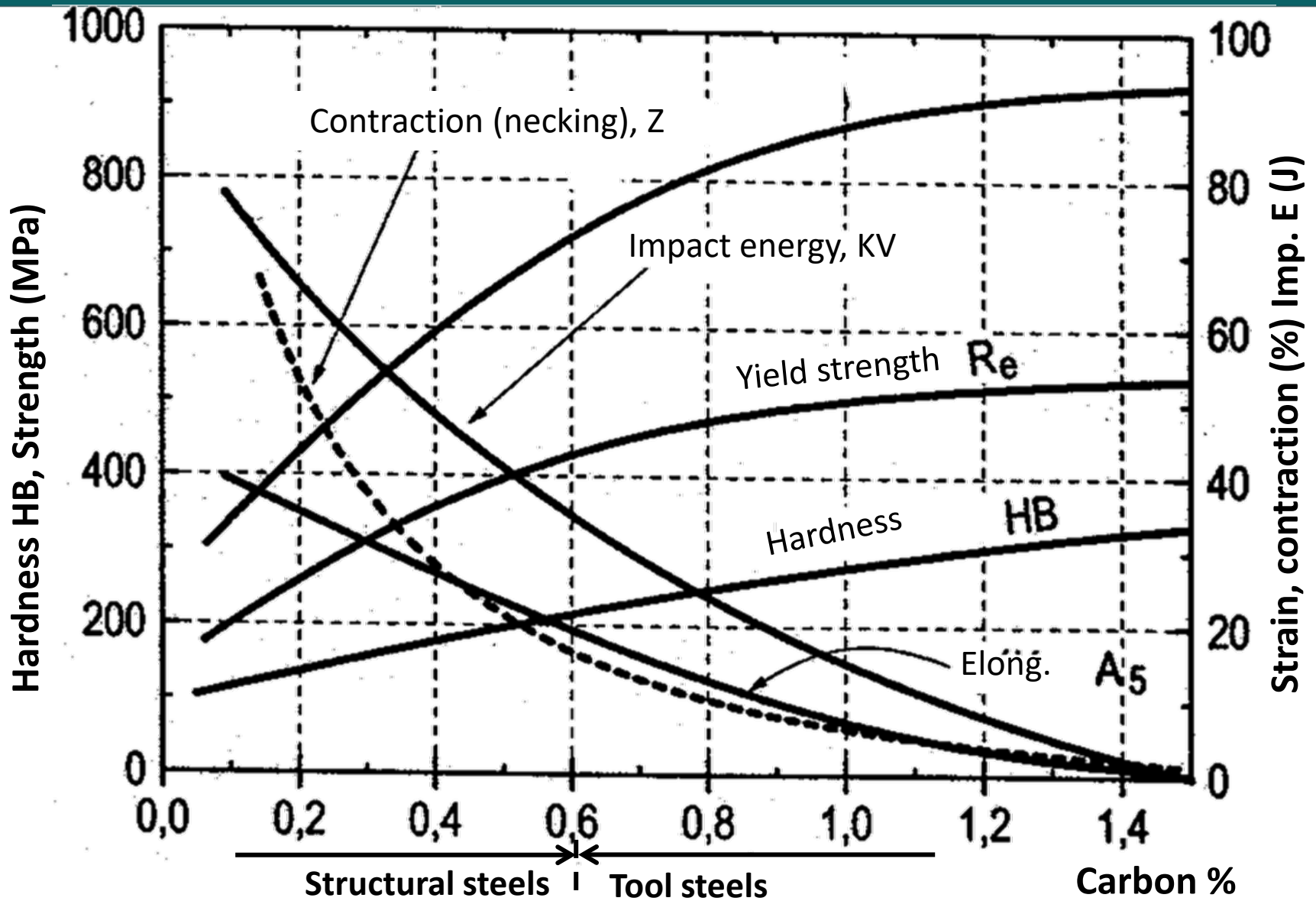
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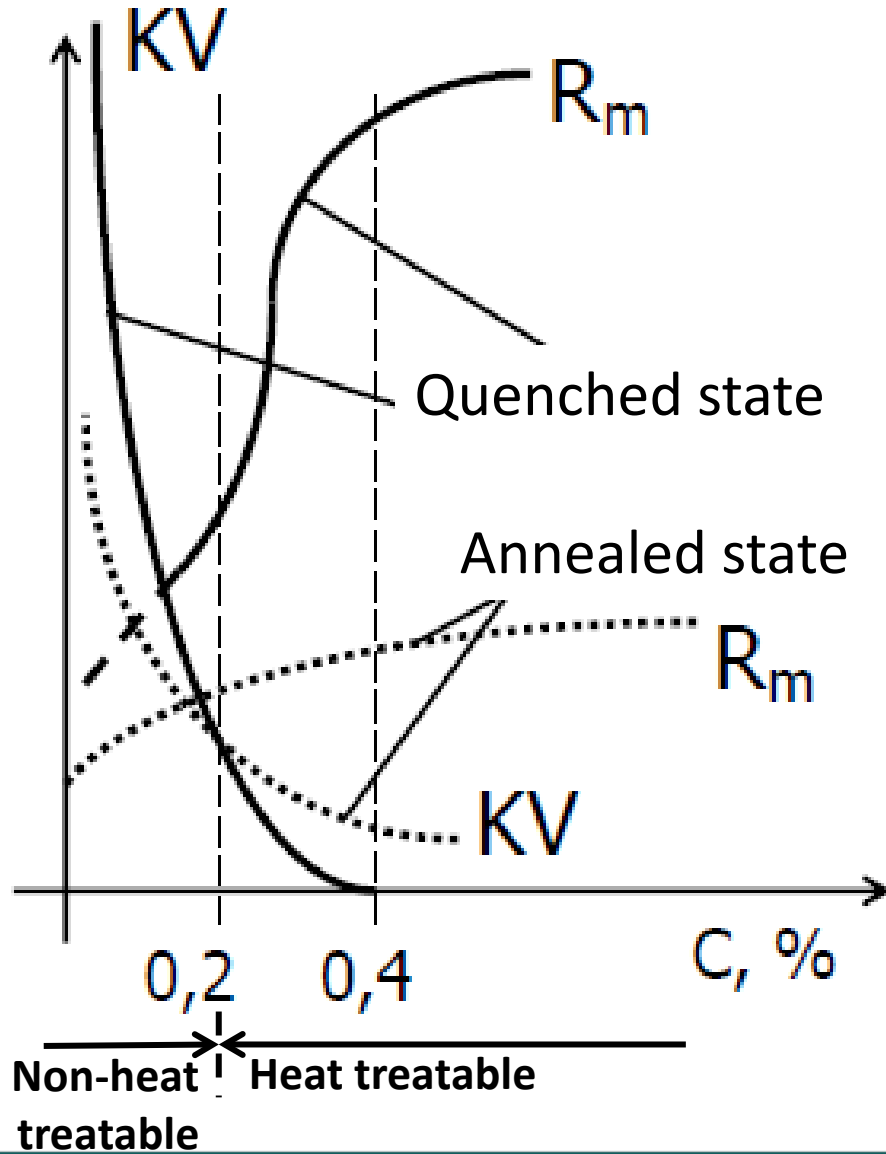
Materials Engineering

BMEGEMTBGF1

2022 Fall semester

- Basic alloying elements
 - C – primer alloying element in most of the steels and cast irons
 - Mn – solid-solution strengthening, deoxidizer, weak austenite promoting element
 - Si – deoxidizer, ferrite promoting element
 - + Ni, Mo, Co, Cr, W, Cu
 - + Al, Ti, V, Zr, B, Ce, Ca, Nb – microalloys
- Impurity elements
 - S – brittleness, sulfides
 - P – brittleness at high temperatures
 - O, H, N – brittleness, ageing, gas porosities
 - + As, Sb, Se, Bi, Sn, Pb





<http://www.indianagroup.com/fabricated-steel-structures/>



Steel structures

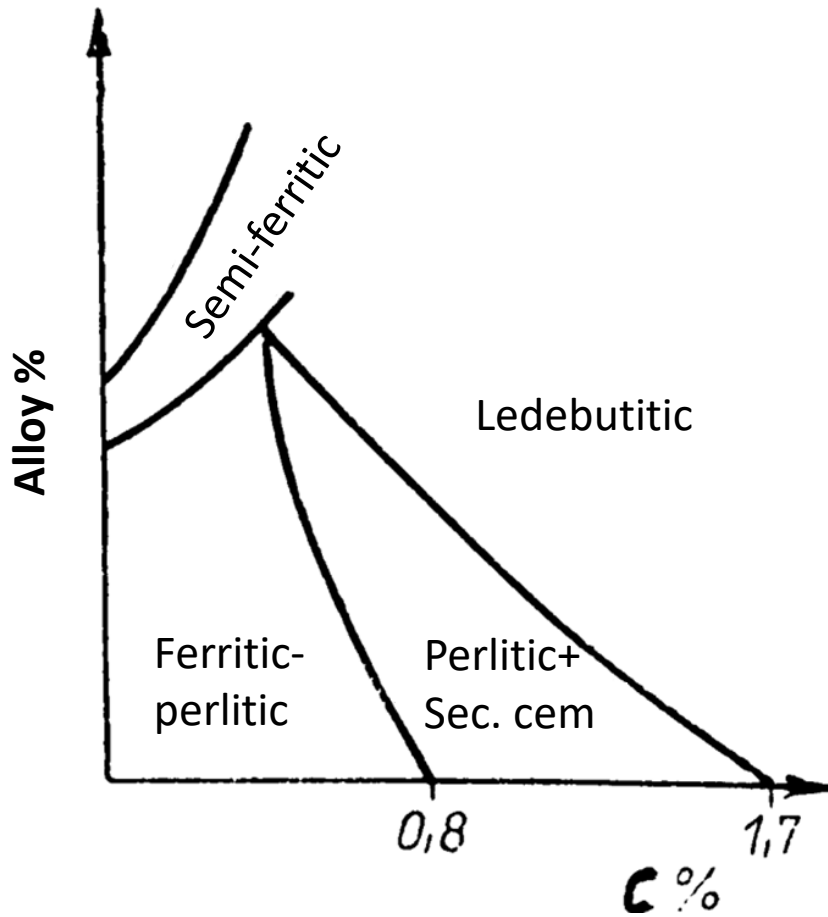
1. Solubility \rightarrow ferrite or austenite producing elements
2. Non-equilibrium $\gamma \rightarrow \alpha$ transformation
3. Austenite grain growth
4. Softening during tempering
5. Embrittlement during tempering
6. Ductile-brittle transition temperature
7. Recrystallization's temperature

Does not dissolve

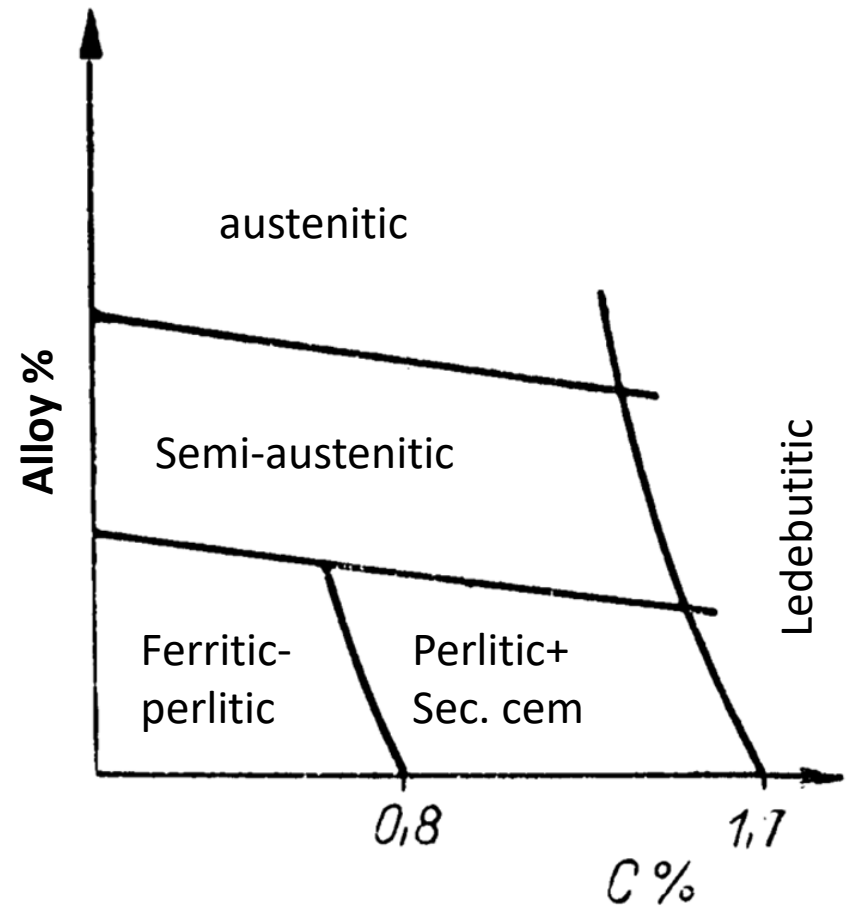
- Produces inclusions, disadvantageous
- S, As, Pb...

Dissolves

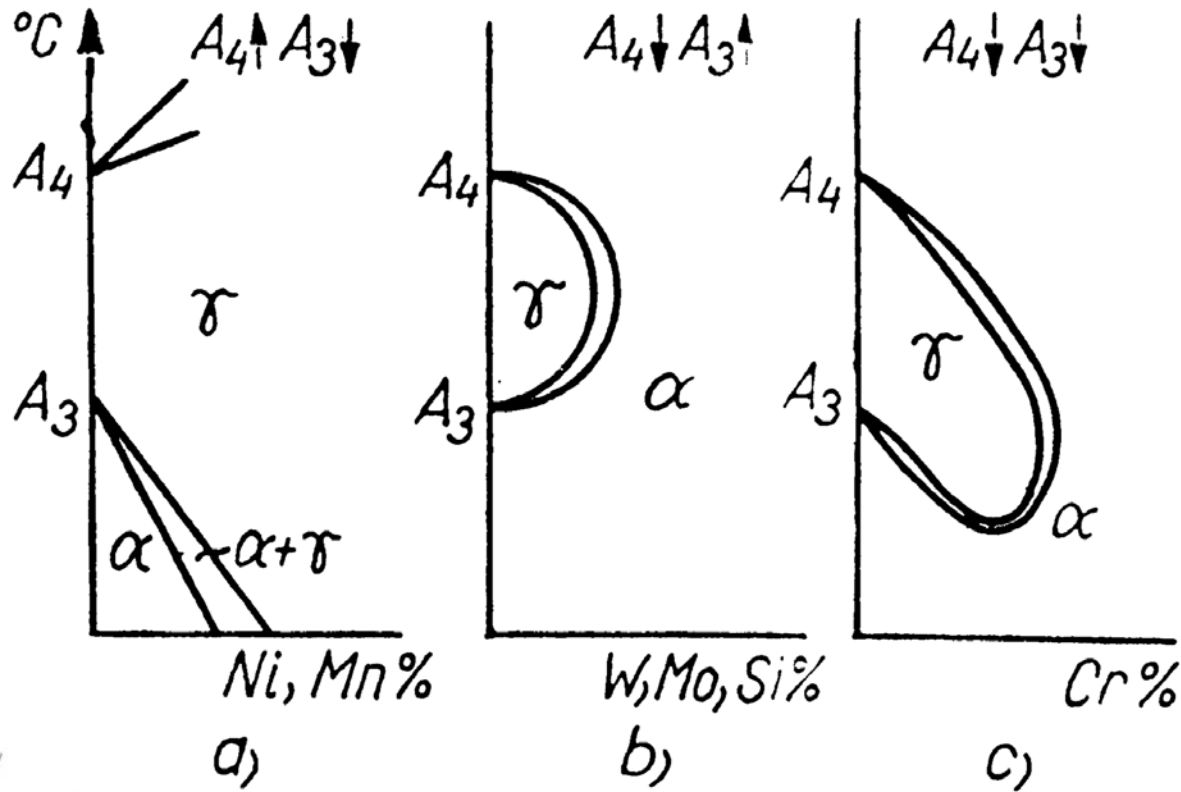
- Dissolves better in ferrite – ferrite promoting element
- Cr, Al, Si, W, Mo, V, Ti
- Dissolves better in austenite – austenite promoting element
- Ni, Mn, C, N, Cu



Ferrite promoting element



Austenite promoting element

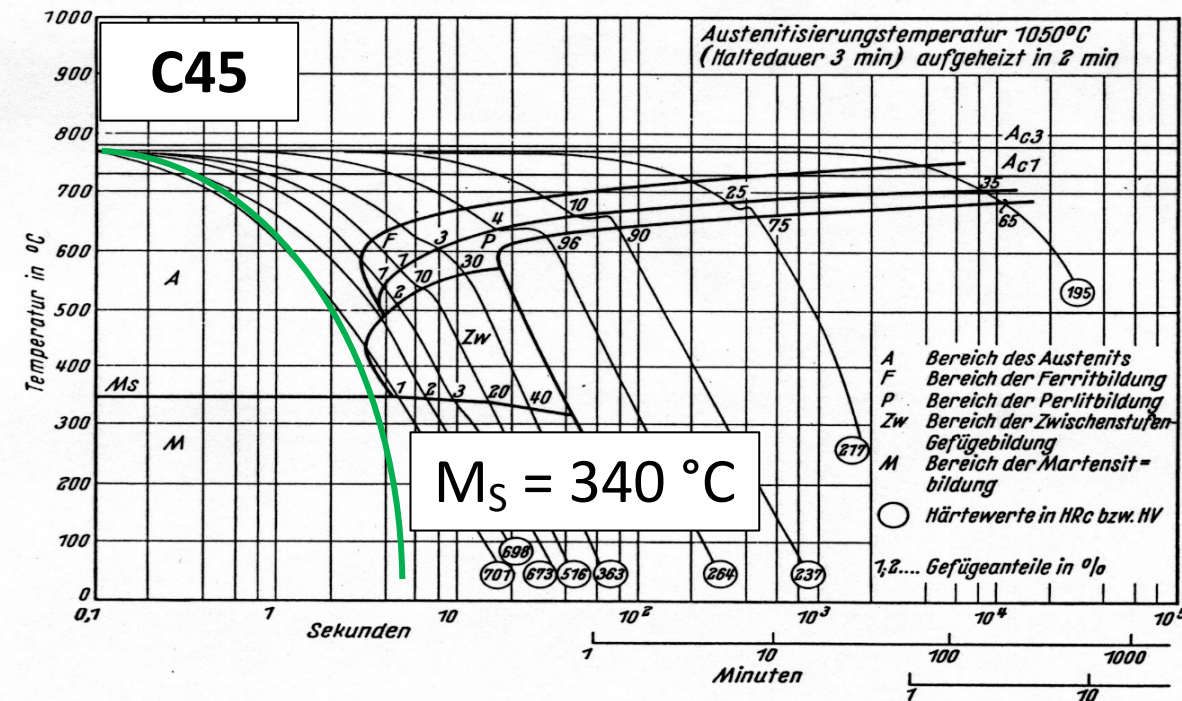
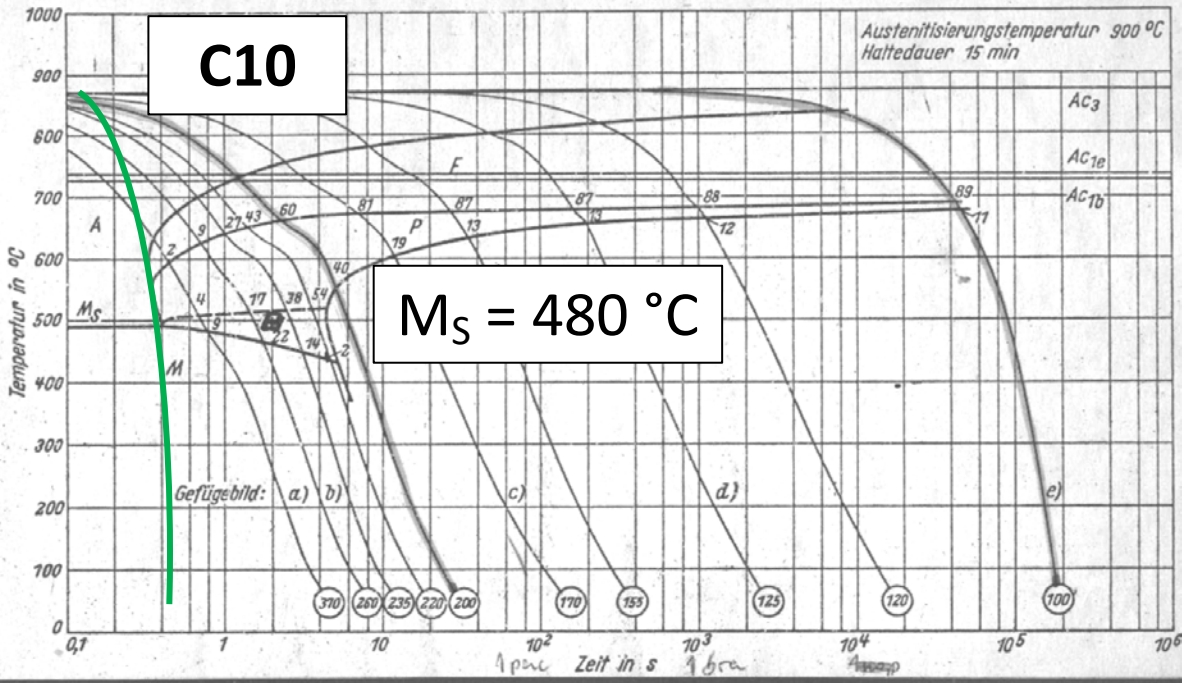
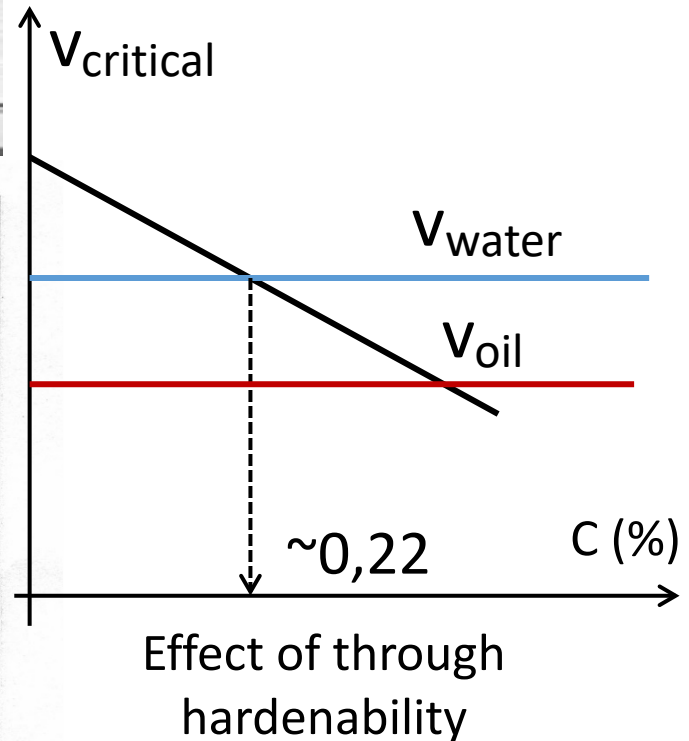


- All alloying elements decreases the M_s and M_f temperatures, except Co and Al.
- The present of residual austenite increases.
 - Deep cooling if necessary
- The CCT curves are shifted to the right.
 - The critical cooling rate is decreasing.
- Hardenability, through hardening diameter increases.

- Importance of quenching: with quenching & tempering (allotropic transformation) the properties can be influenced in wide range.
- Conditions
 - Heating to the temperature of $A_3 + \sim 50^\circ\text{C}$
 - Keeping at constant temperature till material is fully austenitized
 - Cooling faster than the critical cooling rate
 - Practical condition: $C > 0.2\%$

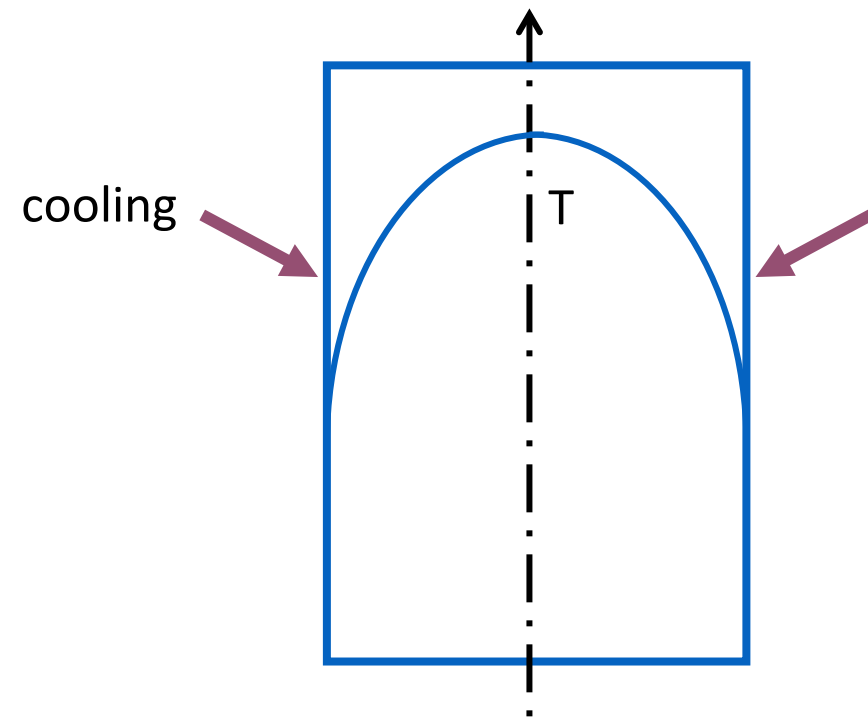
Practical conditions

The alloying decreases the critical cooling rate and the M_s temperature.

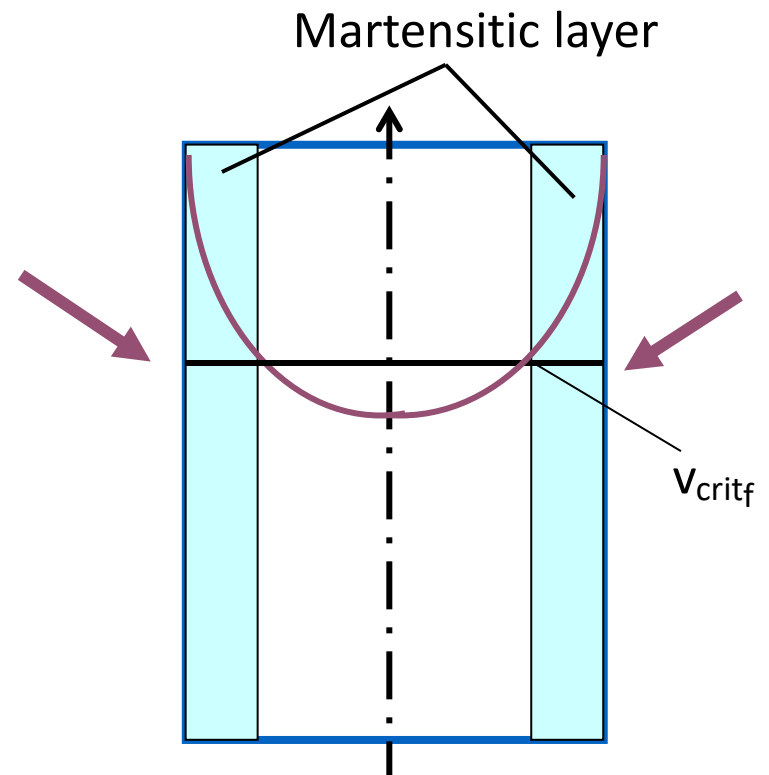


The maximal diameter of a bar, which can be quenched to contain 50 % of martensite.

(see more: lab practice...)

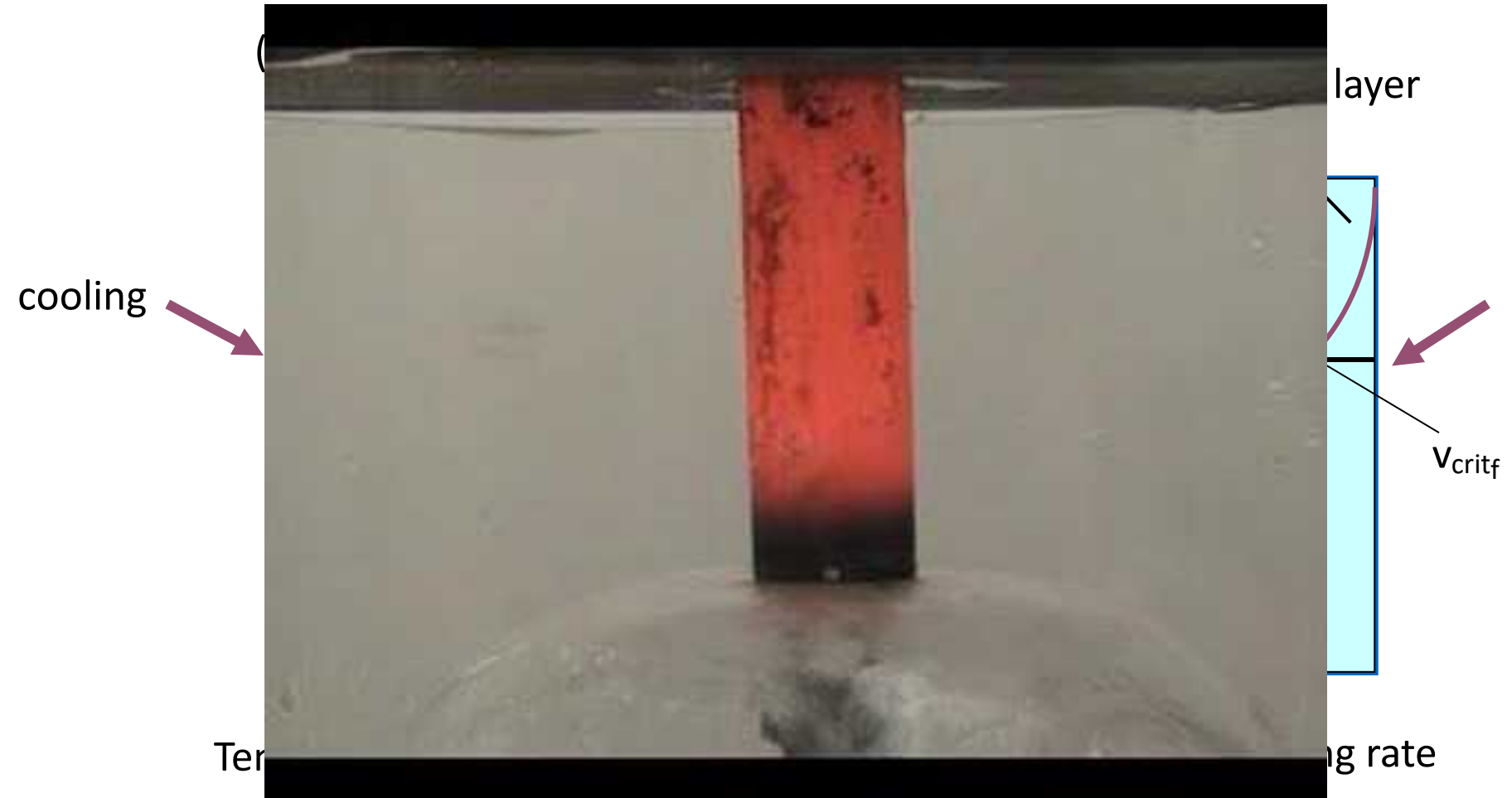


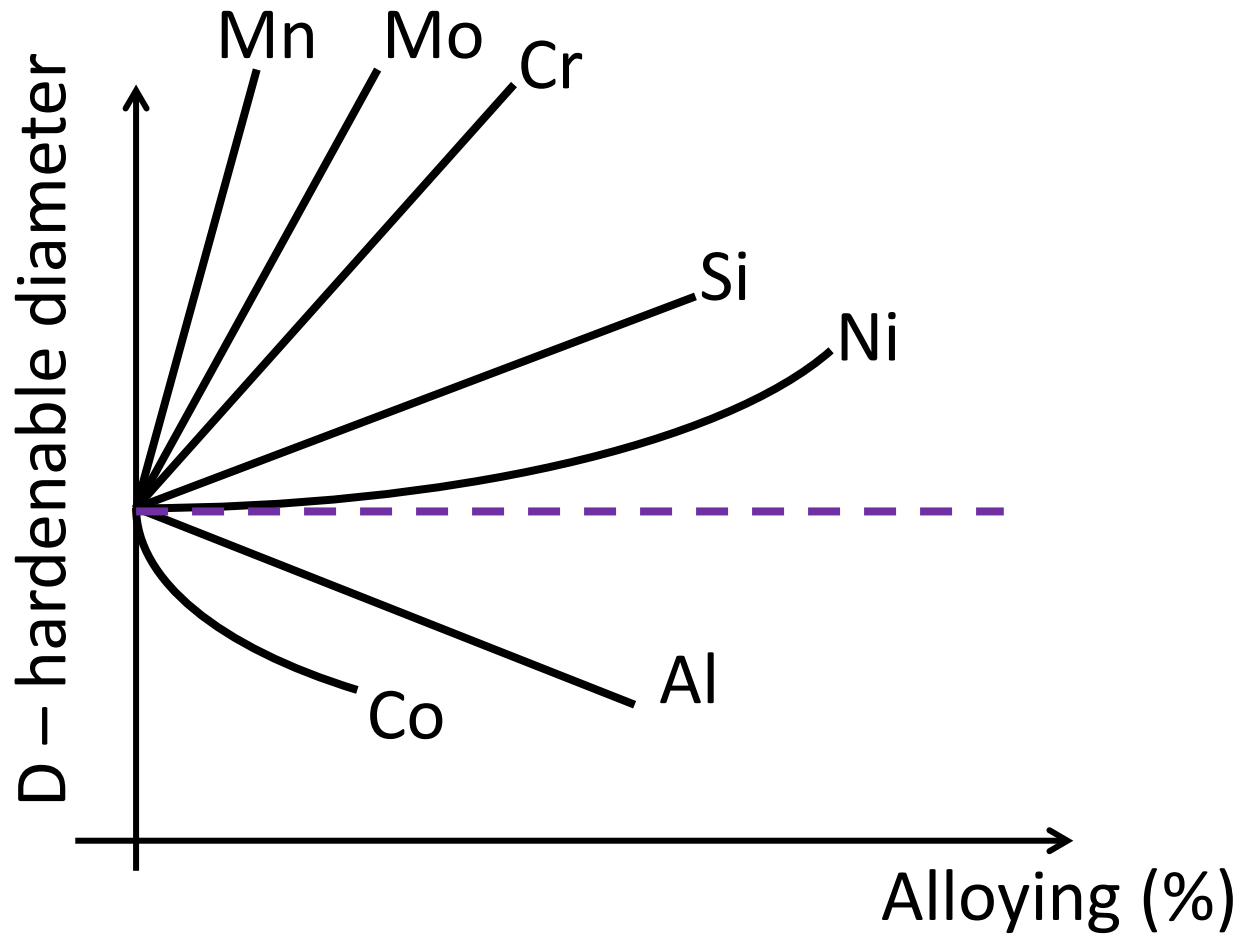
Temperature distribution



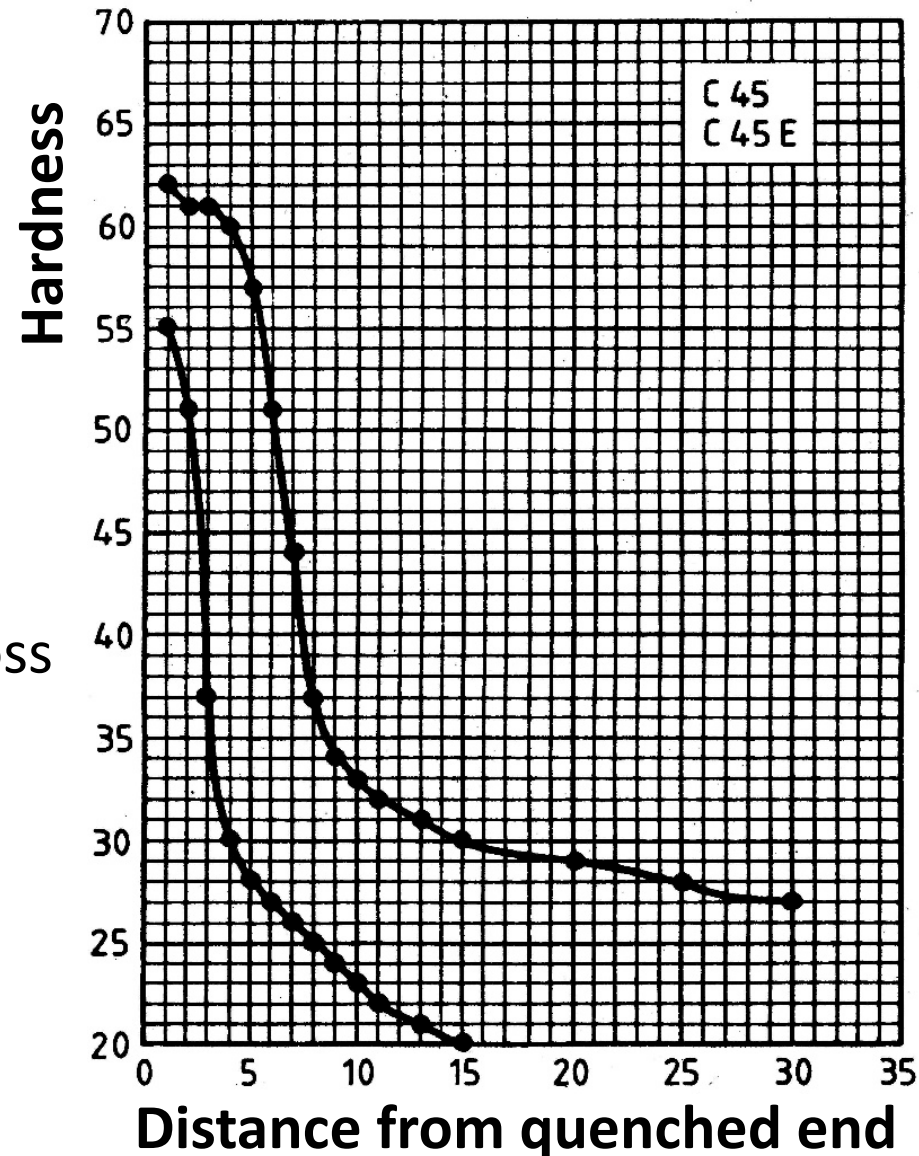
Distribution of cooling rate

The maximal diameter of a bar, which can be quenched to contain 50 % of martensite.



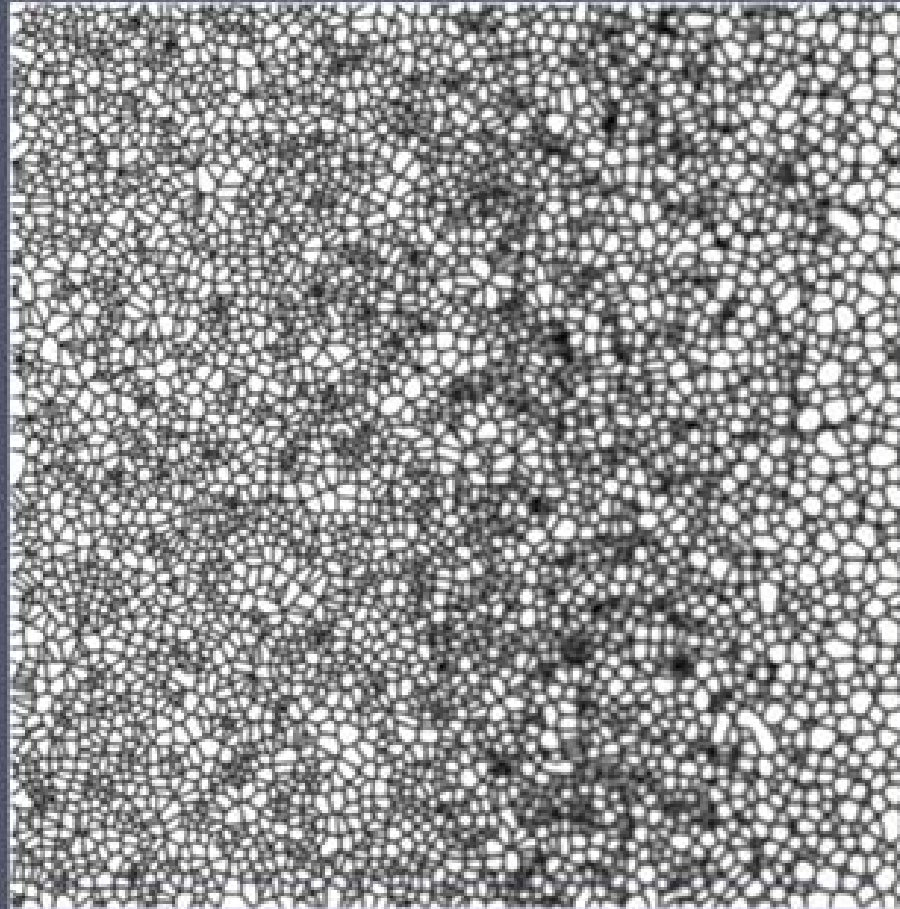


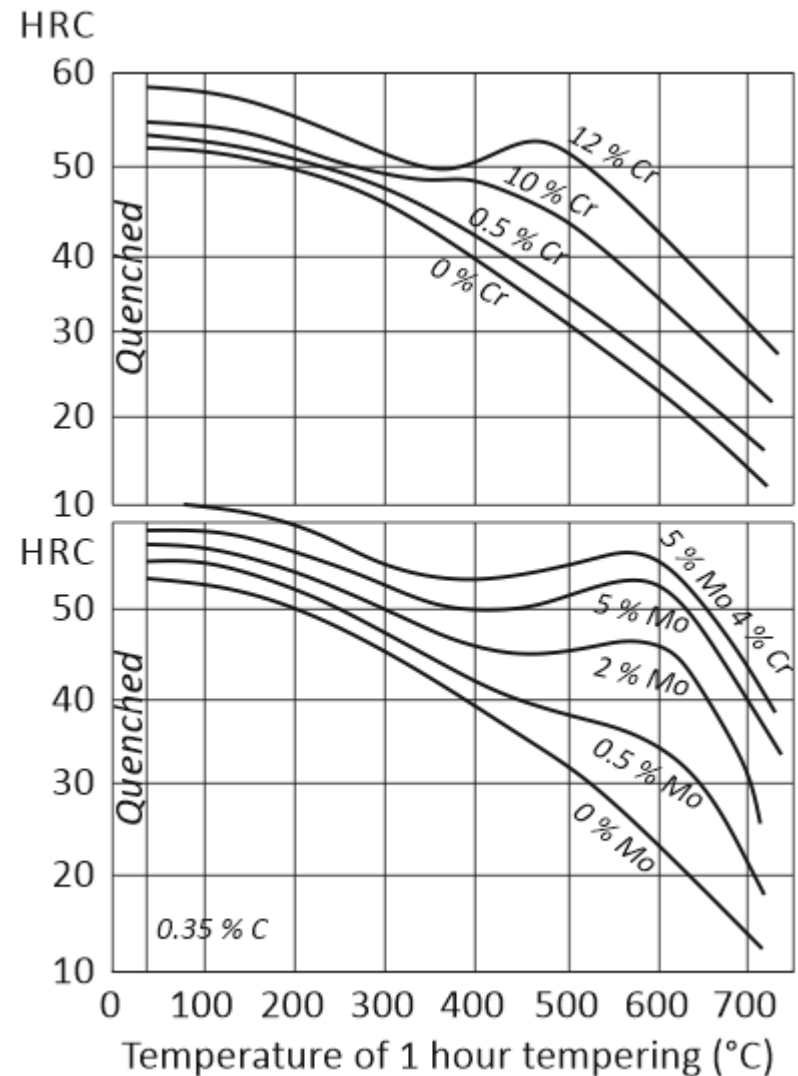
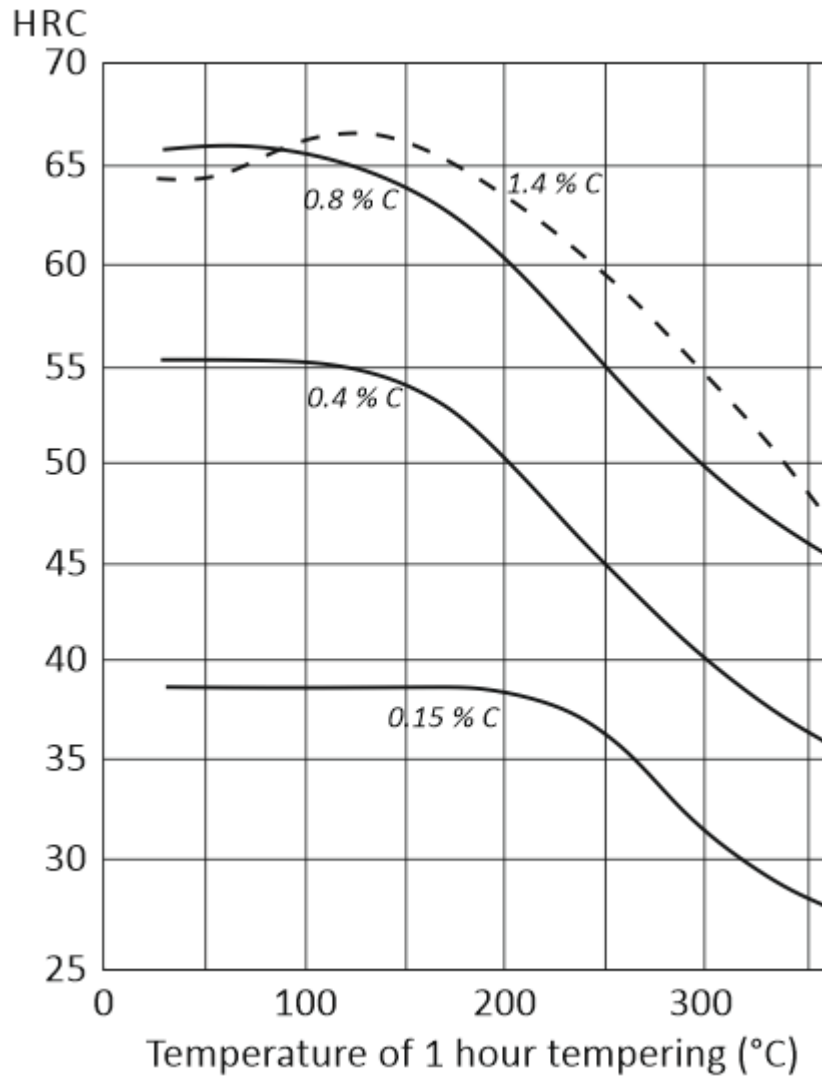
- Verification of material
 - Hardness according to the standards
- Technology information
 - Maximal/minimal hardness by quenching
 - Hardness distribution in the cross section

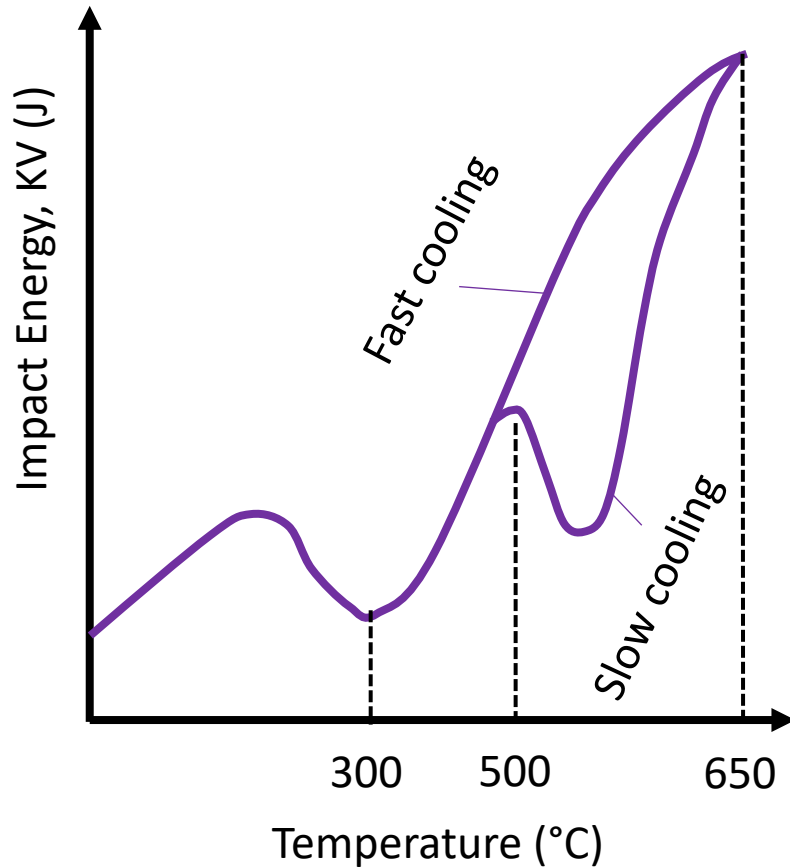


- Mn, Si and B increases the susceptibility to grain coarsening
- Grain refining effect: Ti, V, Nb, Al, Zr
 - Producing fine uniformly distributed nitro-carbides on the grain boundaries, what decreases the boundary migration.
- Other alloying elements have no significant effect of grain coarsening.

3. The effect of alloying on the austenite grain growth



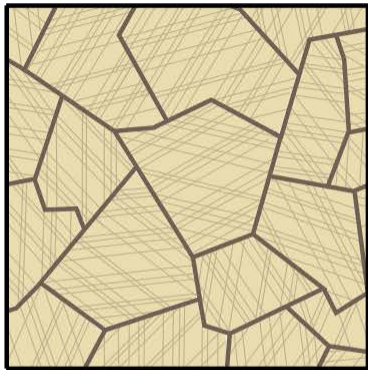




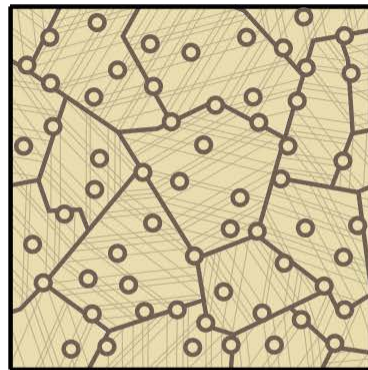
- Cr, Mn causes brittleness if slowly cooled at 500-650 °C
- Reason: Enrichment of carbides, nitrides, phosphides at grain boundaries
- P makes it worse.
- Ni together with Cr and Mn is disadvantageous
- 0.2...0.3 % Mo or 0.5-0.7 % W and fast cooling is advantageous.

- Ni alloying shifts the impact energy–temp. diagram to the left.
 - 1% Ni alloying $\sim 20^{\circ}\text{C}$ shift
- Grain refinement helps as well
 - Nb, V, Ti, Al, Zr, N microalloying $\sim 40^{\circ}\text{C}$ effect
- Impact energy–temp. diagram is shifted to right (makes it worse)
 - C, 0.1% C $\sim 25^{\circ}\text{C}$
 - P, 0.1% P $\sim 55^{\circ}\text{C}$
 - N, 0.01% N $\sim 300^{\circ}\text{C}$ (as solution)
 - O, 0.01% O $\sim 200^{\circ}\text{C}$ (as solution)

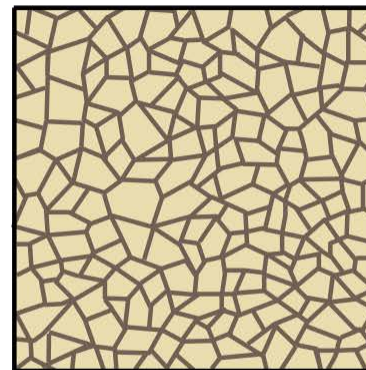
- The alloying increases the heat and creep resistance.
 - W, Mo $\sim 110^\circ\text{C} / \text{at}\%$
 - V $\sim 55^\circ\text{C} / \text{at}\%$
 - Cr $\sim 30^\circ\text{C} / \text{at}\%$



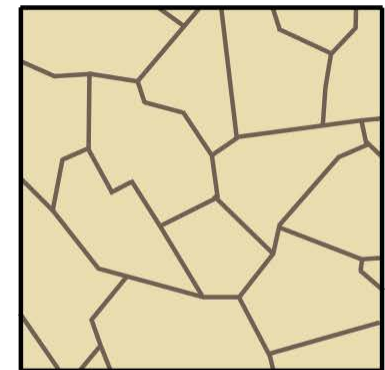
(a)



(b)



(c)



(d)



Thank you for your attention!